

A_{Ω}	Absorptance, directional	-	E	Effect ^a	-
A	Amplitude	varies	\vec{E}	Electric field strength	$\text{V m}^{-1}, \text{N C}^{-1}$
A	Area	m^2	\mathcal{E}	Electromotive force	V
A_{σ}	Area, cross-sectional	m^2	E	Emissions ^g	kg
\dot{A}	Area, interfacial, per unit volume	m^{-1}	\dot{E}	Emissions rate ^g	kg h^{-1}
A_S	Area, surface	m^2	E	Energy	J
A	Arrhenius pre-exponential factor	varies	E_a	Energy, activation	J mol^{-1}
A	Helmholtz energy	J	E_k	Energy, kinetic	J
\vec{a}	Acceleration	m s^{-2}	\dot{E}_k	Energy, kinetic, transfer rate	J s^{-1}
\vec{a}_r	Acceleration, radial	m s^{-2}	E_p	Energy, potential	J
a	Activity	-	\dot{E}_p	Energy, potential, transfer rate	J s^{-1}
a	Turns ratio, primary:secondary ^e	-	\mathcal{E}'	Generator internal voltage, constant ^e	V
a	Van der Waals cohesion pressure	$\text{J m}^3 \text{mol}^{-2}$	E_e	Irradiance	W m^{-2}
B	Bottoms mass flow rate ^d	kg s^{-1}	\mathcal{E}°	Standard electrode potential	V
\vec{B}	Magnetic field flux density	T	E	Young's modulus ^f	Pa
B	Maxwell-Stefan diffusivity, inverted	s m^{-2}	e	Error signal ^b	varies
B_i	McMillan-Mayer virial coefficient i	-	e_s	Error, standard ^a	varies
B_s	Mold constant, Chvorinov	s m^{-2}	e	Residual ^a	varies
B_f	Spectral radiance per unit frequency	$\text{W sr}^{-1} \text{m}^{-2} \text{Hz}^{-1}$	F_0	F-factor ^a	-
B	Susceptance	S	F_c	F-distribution critical value ^a	-
Bi	Biot number	-	F	Feed mass flow rate ^d	kg s^{-1}
BCF	Bioconcentration factor ^g	-	F	Feed molar flow rate	mol s^{-1}
b	Molality	mol kg^{-1}	F_{LV}	Flow parameter ^d	-
b	Steady state value ^b	varies	\vec{F}	Force	N
b	Van der Waals excluded volume	$\text{m}^3 \text{mol}^{-1}$	\vec{F}_b	Force, buoyancy	N
C	Capacitance	F	\vec{F}_d	Force, drag	N
$C_{sb,f}$	Capacity factor ^d	-	\hat{F}_k	Force, friction, kinetic	N
C	Centroid ^f	m	\hat{F}_r	Force, friction, rolling	N
C_0	Coefficient, discharge ^g	-	\hat{F}_s	Force, friction, static	N
C_d	Coefficient, drag	-	\vec{F}	Force, magnetomotive	A-turn
C_P	Coefficient, power	-	\vec{F}_n	Force, normal	N
C_v	Coefficient, valve ^d	-	\vec{F}_r	Force, resultant	N
C	Concentration, mass	kg m^{-3}	\vec{F}_t	Force, tension	N
C	Constant, numerical	-	\hat{F}	Frictional loss	J kg^{-1}
C	Contrast ^a	-	FR	Fractional recovery ^d	-
C_P	Heat capacity, constant pressure	J K^{-1}	Fr	Froude number	-
C_V	Heat capacity, constant volume	J K^{-1}	f_c	Constriction factor	-
\hat{C}	Heat capacity, specific	$\text{J K}^{-1} \text{kg}^{-1}$	f_E	Emissions factor ^g	-
C_H	Heat, humid	$\text{J K}^{-1} \text{kg}^{-1}$	f	Fanning friction factor	-
CI	Confidence interval ^a	varies	f	Fraction / factor	-
Cp	Process capability ^c	varies	f_I	Fraction of inhibition	-
CoP	Coefficient of performance	-	f	Frequency	s^{-1}
c	Concentration, molar	mol m^{-3}	f_n	Frequency, n -th harmonic	s^{-1}
c	Speed of light in medium	m s^{-1}	f°	Fugacity, ideal	Pa
\tilde{c}	Speed of sound in medium	m s^{-1}	f	Fugacity, mixture	Pa
\check{D}	Diffusion per unit volume of reactor	$\text{mol m}^{-3} \text{s}^{-1}$	f^*	Fugacity, pure substance	Pa
D_{AB}	Diffusivity, A through B	$\text{m}^2 \text{s}^{-1}$	$f^{*\ominus}$	Fugacity, pure substance at reference	Pa
\bar{D}	Diffusivity, overall	$\text{m}^2 \text{s}^{-1}$	f_T	Heat exchanger correction factor	-
D_K	Diffusivity, Knudsen	$\text{m}^2 \text{s}^{-1}$	f_Q	Quality factor	-
\mathcal{D}	Diffusivity, Maxwell-Stefan	m s^{-2}	f_s	Scale factor	-
D_a	Dispersion coefficient, axial	$\text{m}^2 \text{s}^{-1}$	f	Vapour fraction of feed ^d	-
D	Distillate mass flow rate ^d	kg s^{-1}	G	Conductance	S
\dot{D}	Drying rate	$\text{kg h}^{-1} \text{m}^{-2}$	G	Gas mass flow rate	kg s^{-1}
\dot{D}_c	Drying rate, constant	$\text{kg h}^{-1} \text{m}^{-2}$	G	Gibbs energy	J
Da	Damköler number	-	\bar{G}	Gibbs energy, partial molar	J mol^{-1}
DR	Decay ratio	-	\vec{G}	Mass velocity	$\text{kg m}^{-2} \text{s}^{-1}$
d	Dilution factor	-	G	Shear modulus ^f	Pa
\vec{d}	Generalised driving force	m^{-1}	Gr	Grashoff number	-
d_2	Unbiasing constant, expected range ^c	-	g_R	Relative centrifugal force	-
d_3	Unbiasing constant, d_2 std. dev. ^c	-	H	Enthalpy	J

\bar{H}	Enthalpy, partial molar	J mol^{-1}	K_w	Water autoionisation constant	-
H	Enthalpy, vapour (<i>cf. h</i>)	J	k	Angular wavenumber	rad m^{-1}
H	Entropy, Shannon	bit	k	Coupling coefficient ^e	-
\tilde{H}	Head	m	k_c	Mass transfer coeff., conc.	m s^{-1}
\tilde{H}_P	Head, pressure	m	k'_c	Mass transfer coeff., conc., EMCD	m s^{-1}
\tilde{H}_v	Head, velocity	m	k_{cy}	Mass transfer coeff., gas, frac.	$\text{mol s}^{-1} \text{m}^{-2} \text{x}^{-1}$
H_H	Heating value, higher	J	k'_y	Mass transfer coeff., gas, frac., EMCD	$\text{mol s}^{-1} \text{m}^{-2} \text{x}^{-1}$
H_L	Heating value, lower	J	k_G	Mass transfer coeff., gas, press.	$\text{mol s}^{-1} \text{m}^{-2} \text{Pa}^{-1}$
\mathcal{H}	Henry's law constant	varies	k'_G	Mass transfer coeff., gas, press., EMCD	$\text{mol s}^{-1} \text{m}^{-2} \text{Pa}^{-1}$
H	Humidity, absolute	$\text{kg}_{\text{water(v)}} \text{kg}_{\text{dry air}}^{-1}$	k_L	Mass transfer coeff., liq., conc.	m s^{-1}
H_c	Humidity, critical	$\text{kg}_{\text{water(v)}} \text{kg}_{\text{dry air}}^{-1}$	k'_L	Mass transfer coeff., liq., conc., EMCD	m s^{-1}
H_M	Humidity, molal	$\text{mol}_{\text{water(v)}} \text{mol}_{\text{dry air}}^{-1}$	k_x	Mass transfer coeff., liq., frac.	$\text{mol s}^{-1} \text{m}^{-2} \text{x}^{-1}$
H_P	Humidity, percentage	-	k'_x	Mass transfer coeff., liq., frac., EMCD	$\text{mol s}^{-1} \text{m}^{-2} \text{x}^{-1}$
H_R	Humidity, relative	-	k_e	Power loss emp. constant, eddy, specific ^e	$\text{W kg}^{-1} \text{Hz}^{-2} \text{T}^{-2} \text{m}^{-2}$
H_S	Humidity, wet bulb temperature	$\text{kg}_{\text{water(v)}} \text{kg}_{\text{dry air}}^{-1}$	k_h	Power loss emp. const., hysteresis, specific ^e	$\text{W kg}^{-1} \text{Hz}^{-2} \text{T}^{-n_s}$
\vec{H}	Magnetic field strength	A m^{-1}	k	Probit parameter ^a	varies
h	Enthalpy, liquid (<i>cf. H</i>)	J	k	Radius of gyration ^f	m
h_d	Fouling coefficient	$\text{W m}^{-2} \text{K}^{-1}$	k	Reaction rate constant	varies
h	Heat transfer coefficient	$\text{W m}^{-2} \text{K}^{-1}$	k_{ad}	Reaction rate constant, absorption	$\text{Pa}^{-1} \text{s}^{-1}$
h	Height	m	k_{de}	Reaction rate constant, desorption	s^{-1}
h_G	Height, gas enthalpy transfer unit ^d	m	\hat{k}_p	Reaction rate constant, pellet, areal	varies
I	Electric current	A	\hat{k}_p	Reaction rate constant, pellet, specific	varies
\dot{I}	Electric current, per unit	pu	\hat{k}_p	Reaction rate constant, pellet, volumetric	varies
\bar{I}	Electric current, phasor domain	A	k_s	Reaction rate constant, surface	varies
I_e	Electron current	s^{-1}	k	Spring constant	N m^{-1}
I	Intensity, sound	W m^{-2}	k	Thermal conductivity	$\text{W m}^{-1} \text{K}^{-1}$
I_{Sx}	Moment of area (2 nd) / area mt. of inertia ^f	m^4	k	Thermal conductivity, effective	$\text{W m}^{-1} \text{K}^{-1}$
I_{Mx}	Moment of mass (2 nd) / mass mt. of inertia ^f	kg m^2	k_{eff}	Thermal conductivity, effective	$\text{W m}^{-1} \text{K}^{-1}$
I_{Pxy}	Product moment of inertia ^f	m^4	\vec{L}	Angular momentum	$\text{kg m}^2 \text{s}^{-1}$
i	Electric current, time dependent	A	L	Inductance	H
i	van 't Hoff factor	-	L_m	Inductance, mutual	H
J_D	Colburn mass transfer factor	-	L	Length, dimensionless	-
J	Current density	A m^{-2}	L	Liquid mass flow rate ^d	kg s^{-1}
\vec{J}	Impulse	N s	\bar{L}	Liquid mass flow rate in reboiler ^d	kg s^{-1}
J	Ising model interaction	-	L_S	Solid loading	kg
K_a	Acid dissociation constant	-	\dot{L}_S	Solid loading mass flow rate	kg s^{-1}
K_b	Base dissociation constant	-	L_I	Sound intensity level ^g	dB
K	Bulk modulus ^f	Pa	LCL	Lower control limit ^c	varies
K_{fp}	Cryoscopic constant	kg K mol^{-1}	LFL	Lower flammability limit ^g	%
K_d	Dry soil-water partition coefficient ^g	-	LOC	Limiting oxygen concentration ^g	%
K_{bp}	Ebullioscopic constant	kg K mol^{-1}	LSL	Lower specification limit ^c	varies
K_e	Equilibrium constant	-	ℓ	Azimuthal quantum number	-
K_{ad}	Equilibrium constant, adsorption	Pa^{-1}	ℓ	Length	m
K_{cat}	Equilibrium constant, catalysis	-	ℓ_a	Length, arc	m
K_c	Equilibrium constant, concentration	-	ℓ'	Length, characteristic	m
K_{de}	Equilibrium constant, desorption	Pa	ℓ_f	Length, focal	m
K_p	Equilibrium constant, pressure	-	M	Conductor material temperature constant ^e	K
K_s	Equilibrium constant, surface reaction	varies	M	Modulus	m
K'	Equilibrium distribution coefficient ^d	-	M_T	Modulus, Thiele	-
K	Equilibrium vapour-liquid ratio ^d	-	M	Molarity	mol L^{-1}
K_f	Fitting loss coefficient	-	\vec{M}	Moment, bending ^f	N m
K_v	Flow factor	$\text{m}^3 \text{h}^{-1} \text{bar}^{-1/2}$	M_x	Moment of mass (1 st) ^f	kg m
K_I	Fracture toughness, plane-strain ^f	$\text{Pa m}^{1/2}$	M	Optical magnification	-
K_M	Michaelis constant	mol m^{-3}	MS	Mean square ^a	-
K_{ow}	Octanol-water partition coefficient ^g	-	MSE	Mean squared error ^a	-
K_{oc}	Organics soil-water partition coeff. ^g	-	\mathbf{m}	Flux, mass, relative to $\langle \vec{v}_m \rangle$	$\text{kg s}^{-1} \text{m}^{-2}$
K_p	Process gain ^b	-	\mathbf{m}'	Flux, mass, relative to $\langle \vec{v}_n \rangle$	$\text{kg s}^{-1} \text{m}^{-2}$
K_c	Proportional gain constant ^b	-	m_ℓ	Magnetic quantum number	-
K_{sp}	Solubility product	varies	m	Mass	kg
K_s	System gain ^b	-	m_a	Mass, atomic	Da
			\dot{m}	Mass flow rate	kg s^{-1}

\tilde{m}	Molar mass (molecular weight)	g mol^{-1}	\tilde{Q}	Volumetric heat generation	W m^{-3}
$\langle \tilde{m} \rangle$	Molar mass, average	g mol^{-1}	q_λ	Charge density, linear	C m^{-1}
m	Order parameter, magnetic, per spin	-	q_σ	Charge density, surface	C m^{-2}
N_e	Number of electrons transferred	-	q_ρ	Charge density, volume	C m^{-3}
N_L	Number of levels ^a	-	q	Electric charge	C
N_t	Number of turns in a coil	-	q	Liquid fraction of stream ^d	-
N	Quantity	-	q	Mass ratio of adsorbate to adsorbent ^d	-
\dot{N}_r	Revolutions per minute	min^{-1}	R^2	Coefficient of determination ^a	-
N	Size of population ^a	-	R	Consumption/generation rate, mass	kg s^{-1}
Nu	Nusselt number	-	\tilde{R}	Cons./gen. rate, mass, volumetric	$\text{kg s}^{-1} \text{m}^{-3}$
Nu _p	Nusselt number, particle	-	\hat{R}	Gas constant, specific ^e	$\text{J kg}^{-1} \text{K}^{-1}$
NPSH _A	Net positive suction head available	m	R	Radius	m
n	Diffraction order	-	\bar{R}	Radius, geometric mean ^e	m
n_H	Hill coefficient	-	\bar{R}	Range, average ^c	-
n	Index of refraction	-	R	Reflux ratio ^d	-
\mathbf{n}	Flux, molar, relative to $\langle \vec{v}_m \rangle$	$\text{mol s}^{-1} \text{m}^{-2}$	\mathcal{R}	Reluctance	H^{-1}
\mathbf{n}'	Flux, molar, relative to $\langle \vec{v}_n \rangle$	$\text{mol s}^{-1} \text{m}^{-2}$	R_f	Resistance of filter medium ^d	m^{-1}
\dot{n}	Molar flow rate	mol s^{-1}	R	Resistance	Ω
n	Number of moles	mol	\dot{R}	Resistance, per unit	pu
n	Number of samples ^a	-	R	Resistance, thermal	K W^{-1}
n	Polytropic index	-	R	Tunneling reflection coefficient	-
n	Principal quantum number	-	Re	Reynolds number	-
n	Size of sample ^a	-	Re _p	Reynolds number, particle	-
n_s	Steinmetz coefficient ^e	-	r_b	Boilup ratio ^d	-
OSFC	Out of service fuel concentration ^g	%	r	Consumption/generation rate, molar	mol s^{-1}
\vec{P}	Momentum	kg m s^{-1}	\check{r}	Cons./gen. rate, molar, volumetric	$\text{mol s}^{-1} \text{m}^{-3}$
P^*	Overpressure, scaled ^g	-	r	Position, 2D space	m
P_{gs}	Permeability, gas-solid ^d	$\text{m}^3_{\text{solute}} \text{s}^{-1} \text{m}^{-2} \text{Pa}^{-1} \text{m}_\delta$	\vec{r}	Position vector, 2D space	m
P_G	Permeability, membrane, gas ^d	$\text{mol s}^{-1} \text{m}^{-1} \text{Pa}^{-1}$	r	Radial coordinate position	m
P_L	Permeability, membrane, liquid ^d	m s^{-1}	r	Rate of reaction	varies
P_{slv}	Permeability, membrane, solvent ^d	$\text{kg}_{\text{solvent}} \text{s}^{-1} \text{m}^{-1} \text{Pa}^{-1}$	\hat{r}	Rate of reaction, catalysed, specific	$\text{mol}_{\text{cat}}^{-1} \text{s}^{-1}$
\dot{P}_e	Power loss, eddy, specific	W kg^{-1}	\check{r}	Rate of reaction, catalysed, volumetric	$\text{mol L}^{-1} \text{s}^{-1}$
\dot{P}_h	Power loss, hysteresis, specific	W kg^{-1}	r	Ratio	-
P	Power, real	W	S	Entropy	J K^{-1}
\dot{P}	Power, real, per unit	pu	\bar{S}	Entropy, partial molar	$\text{J mol}^{-1} \text{K}^{-1}$
\mathcal{P}	Poynting correction factor	-	S	Moment of area (1 st) ^f	m^3
P	Pressure	Pa	S	Power, apparent complex	V A
P_c	Pressure, critical	Pa	\dot{S}	Power, apparent complex, per unit	pu
\mathcal{P}	Pressure, dynamic	Pa	\vec{S}	Poynting vector	W m^{-2}
PB	Proportional band ^b	-	\mathcal{S}	Saturation, absolute	$\text{kg}_{(v)} \text{kg}_{\text{dry gas}}^{-1}$
Pe	Péclet number	-	\mathcal{S}_M	Saturation, molal	$\text{mol}_{(v)} \text{mol}_{\text{dry gas}}^{-1}$
PI	Prediction interval ^a	varies	\mathcal{S}_p	Saturation, percentage	-
PO	Percent overshoot ^b	-	\mathcal{S}_R	Saturation, relative	-
Pr	Prandtl number	-	$\mathcal{S}_{A/B}$	Selectivity of A to B	-
\vec{p}	Electric dipole moment	C m	$\bar{\mathcal{S}}_{A/B}$	Selectivity of A to B, overall	-
p	Power, instantaneous	W	\vec{S}	Spin angular momentum	$\text{m}^2 \text{kg s}^{-1}$
p	Pressure, partial	Pa	S	Steam mass flow rate	kg s^{-1}
p^*	Pressure, vapour	Pa	$\langle \vec{S} \rangle$	Time-averaged Poynting vector	W m^{-2}
p	Probability value ^a	-	Sc	Schmidt number	-
p	System pole ^b	-	Sh	Sherwood number	-
pf	Power factor	-	Sh _p	Sherwood number, particle	-
Q	Heat	J	SP	Split ^d	-
\dot{Q}	Rate of heat flow	W	SS	Sum of squares ^a	-
\dot{Q}_x	Rate of heat flow in x direction	W	St	Stanton number	-
Q_r	Reaction quotient	-	SIL	Surge impedance load ^e	W
Q_c	Reaction quotient, molar concentration	-	SSE	Sum of squares residual error ^a	-
Q_p	Reaction quotient, pressure	-	SST	Sum of squares, total ^a	-
Q	Reactive complex power	VAR	s	Complex frequency	complex Np s^{-1}
\dot{Q}	Reactive complex power, per unit	pu	s	Position, 1D space	m
Q	Total charge enclosed	C	\vec{s}	Position vector, 1D space	m

s	Solubility, gas	$m^3_{\text{solute}} \ominus m^{-3}_{\text{solid}} \text{Pa}^{-1}$	W	Work	J
s	Spin quantum number	-	\dot{W}	Work rate	W
s	Standard deviation, sample ^a	varies	\hat{W}	Work per unit of mass pumped	J kg^{-1}
s^2	Variance, sample ^a	varies	w	Mass fraction	-
T	Period	s	ω	Width	m
$T_{1/2}$	Period, half	s	$\vec{\omega}$	Vorticity	s^{-1}
T	Temperature, thermodynamic	K	X	Conversion fraction	-
T_c	Temperature, critical	K	X_e	Conversion fraction, equilibrium	-
T_D	Temperature, Debye	K	X	Reactance	Ω
T_r	Temperature, reduced	-	\dot{X}	Reactance, per unit	pu
T_Ω	Transmittance, directional	-	\bar{X}	Reactance, phasor domain	Ω
T	Tunneling transmission coefficient	-	X'	Reactance, transient	Ω
TLV	Threshold limit value ^g	varies	x	Coded level variable ^a	-
TWA	Time-weighted average ^g	varies	x	Mole fraction of more dense species (<i>cf.</i> y)	-
$t_{1/2}$	Half-life	s	\bar{x}	Centre of mass ^f	m
\vec{t}	Stress vector	Pa	Y	Admittance	S
t	Time	s	Y_{ES}	Fractional occupancy of ligand S on receptor E	-
t_m	Time, mean residence	s	Y	Geometric factor for toughness ^f	-
t_p	Time, peak ^b	s	Y	Probit variable ^a	-
t_s	Time, solidification ^f	s	Y_c	Yield, casting	-
U	Energy, internal	J	Y_A	Yield of A	-
\hat{U}	Energy, internal, specific	J kg^{-1}	\bar{Y}_A	Yield of A, overall	-
U_E	Energy, potential, electrical	J	y	Mole fraction of less dense species (<i>cf.</i> x)	-
U_M	Energy, potential, magnetic	J	y	Process variable ^b	varies
U	Overall heat transfer coefficient	$\text{W m}^{-2} \text{K}^{-1}$	\hat{y}	Regression model, fitted ^a	-
U	Ullage ^g	m^3	Z	Atomic number	-
U	Velocity, dimensionless	-	Z	Collision frequency	s^{-1}
UCL	Upper control limit ^c	varies	Z	Compressibility factor	-
UFL	Upper flammability limit ^g	%	Z_c	Comp. factor at critical point	-
USL	Upper specification limit ^c	varies	Z	Impedance	Ω
u_0	Bias ^b	varies	Z_0	Impedance, characteristic	Ω
u	Control signal (manipulated variable) ^b	varies	\dot{Z}	Impedance, per unit	pu
u	Energy density	J m^{-3}	\bar{Z}	Impedance, phasor domain	Ω
u_f	Energy density, spectral	$\text{J m}^{-3} \text{Hz}^{-1}$	Z	Partition function, canonical	-
u_∞	Velocity, freestream	m s^{-1}	Z	Partition function, grand canonical	-
V	Causative variable ^a	varies	z	Fraction in feed ^d	-
V	Vapour mass flow rate ^d	kg s^{-1}	z	Partition function, isothermal-isobaric	-
\bar{V}	Vapour mass flow rate in reboiler ^d	kg s^{-1}	z	Position, z-axis	m
V	Voltage	V, J C^{-1}	z_e	Scaled distance ^g	$\text{m kg}^{-1/3}$
\bar{V}	Voltage, phasor domain	V	z	System zero ^b	-
\dot{V}	Voltage, per unit	pu			
V	Volume	m^3			
V_c	Volume, critical	m^3			
V_H	Volume, humid	$\text{m}^3 \text{kg}_{\text{dry air}}^{-1}$			
\bar{V}	Volume, partial molar	$\text{m}^3 \text{mol}^{-1}$			
\bar{V}_s	Volume, saturated molar	$\text{m}^3 \text{mol}^{-1}$			
\dot{V}	Volumetric flow rate	$\text{m}^3 \text{s}^{-1}$			
v	Atomic/structural diffusion vol. increment	-			
v	Reaction velocity	$\mu\text{mol L}^{-1} \text{min}^{-1}$			
\bar{v}	Reaction velocity, net	$\mu\text{mol L}^{-1} \text{min}^{-1}$			
\vec{v}	Velocity	m s^{-1}			
$\langle \vec{v} \rangle$	Velocity, average	m s^{-1}			
$\langle \vec{v}_m \rangle$	Velocity, average, mass	m s^{-1}			
$\langle \vec{v}_n \rangle$	Velocity, average, molar	m s^{-1}			
\bar{v}_d	Velocity, drift	m s^{-1}			
\bar{v}_ϕ	Velocity, phase	m s^{-1}			
\bar{v}_s	Velocity, superficial	m s^{-1}			
\bar{v}_t	Velocity, tangential	m s^{-1}			
\bar{v}_∞	Velocity, terminal	m s^{-1}			
v	Voltage, time dependent	V			

Greek Alphabet

$\vec{\alpha}$	Acceleration, angular	rad s^{-2}
α	Arrhenius number	-
α	Attenuation constant	Np m^{-1}
α	Cooperativity factor	-
α	Dissociation fraction	-
α°	Ideal separation factor ^d	-
α	Kinetic energy velocity correction factor	-
α	Order of reaction	-
$\bar{\alpha}$	Order of reaction, overall	-
α_{AB}	Relative volatility ^d of A to B	-
α	Significance level ^a	-
α	Specific filter cake resistance ^d	m kg^{-1}
α	Thermal diffusivity	$\text{m}^2 \text{s}^{-1}$
α	Thermal expansion coefficient	K^{-1}
β	Bandwidth ^b	s^{-1}
β	Phase constant	rad m^{-1}
β_0	Regression coefficient, intercept ^a	-
$\hat{\beta}_0$	Regression coefficient, intercept, est. ^a	-
β_1	Regression coefficient, slope ^a	-
$\hat{\beta}_1$	Regression coefficient, slope, est. ^a	-

β	Thermodynamic beta / coldness	J^{-1}	κ	Viscosity, volume	Pa s
Γ	Thermodynamic factor	-	Λ	Grand potential	J
γ	Activity coefficient	-	λ	Decay constant	s^{-1}
γ	Heat capacity ratio	-	λ	Eigenvalue ^b	-
γ	Lorentz factor	-	λ	Flux linkage ^e	Wb-turn
γ_i	Overshoot of peak i above stdy. state ^b	varies	λ	Latent heat of vapourisation	$J kg^{-1}$
γ	Propagation constant ^e	m^{-1}	λ	Mean free path	m
γ_{xy}	Shear strain	-	λ	Tip to speed ratio ^e	-
γ	Specific weight	$N m^{-3}$	λ	Wavelength	m
γ	Surface tension	$N m^{-1}$	μ	Chemical potential	$J mol^{-1}$
Δ_{set}	Steady state offset ^b	varies	μ_k	Coefficient of friction, kinetic	-
$\Delta_r C_P$	Heat capacity of reaction	$J K^{-1}$	μ_r	Coefficient of friction, rolling	-
$\Delta_r G$	Gibbs energy of reaction	J	μ_k	Coefficient of friction, static	-
$\Delta_{sln} H$	Enthalpy of solution	$J mol_{solute}^{-1}$	μ_J	Joule coefficient	$K m^{-3}$
$\Delta_c H$	Heat of combustion	J	μ_{JT}	Joule-Thomson coefficient	$K Pa^{-1}$
$\Delta_f H$	Heat of formation	J	μ	Linear density	$kg m^{-1}$
$\Delta_{fus} H$	Heat of fusion	J	$\vec{\mu}$	Magnetic dipole moment	$J T^{-1}$
$\Delta_{mix} H$	Heat of mixing	J	$\vec{\mu}_S$	Magnetic dipole moment, spin	$J T^{-1}$
$\Delta_r H$	Heat of reaction	J	μ_{AB}	Mass, reduced	kg
$\Delta_{trs} H$	Heat of phase transition	J	μ	Mean, population ^a	varies
$\Delta_{vap} H$	Heat of vapourisation	J	μ	Viscosity, dynamic	Pa s
$\Delta_{mix} S$	Entropy of mixing	$J K^{-1}$	ν	Degrees of freedom	-
$\Delta_{trs} S$	Entropy of phase transition	$J K^{-1}$	ν	Poisson's ratio ^f	-
$\Delta_r S$	Entropy of reaction	$J K^{-1}$	ν	Stoichiometric coefficient	-
Δ_s	Spacetime interval	m	ν	Viscosity, kinematic	$m^2 s^{-1}$
$\Delta_{fus} V$	Volume change of fusion	m^3	ξ	Extent of reaction	mol
$\Delta_{trs} V$	Volume change of phase transition	m^3	$\dot{\xi}$	Extent of reaction, continuous	$mol s^{-1}$
$\Delta_{sh} \rho$	Density change at superheat	$g cm^{-3} \text{ } ^\circ C^{-1}$	Π	Osmotic pressure	Pa
$\bar{\delta}$	Distance, geometric mean ^e	m	π_T	Internal pressure	Pa
δ	Power angle ^e	rad	ρ_H	Density, humid	$kg m^{-3}$
δ	Thickness / distance	m	ρ	Density, mass	$kg m^{-3}$
ε_Ω	Emissivity, directional	-	ρ_N	Density, numerical	m^{-3}
ϵ	Energy quantum	varies	ρ_R	Density, relative (specific gravity)	-
ϵ	Heat exchanger effectiveness	-	ρ	Resistivity	Ωm
ε_f	Non-ideal filling factor ^g	-	ρ_s	Steric factor	-
ε_m	Non-ideal mixing factor ^g	-	σ	Atomic radius, hard elastic sphere	Å
ε	Random error ^a	-	σ_{AB}	Collision diameter (cross section)	Å
ϵ	Rate of strain tensor	s^{-1}	σ	Conductivity	$S m^{-1}$
ε_R	Relative permittivity	-	σ	Density, surface	$g m^{-2}$
ε_e	Strain, engineering ^f	-	σ	Ising spin configuration	-
ε	Strain, true ^f	-	σ	Neper frequency	$Np s^{-1}$
ε	Volumetric expansion parameter	-	$\sigma_{x,y,z}$	Standard deviation, Gaussian dispersion ^a	m
ζ	Damping ratio ^b	-	σ	Standard deviation, population ^a	-
η_{int}	Catalyst internal effectiveness factor	-	$\hat{\sigma}$	Standard deviation, pop., estimated ^a	-
η	Efficiency	-	σ_e	Stress, engineering ^f	Pa
η_{dc}	Frac. of $A_{\sigma,col}$ for vapour flow ^d	-	σ	Stress, true ^f	Pa
η_{ML}	Murphree liquid efficiency ^d	-	σ	Total stress tensor	Pa
η_{MV}	Murphree vapour efficiency ^d	-	σ	Van der Waals radius	Å
Θ_B	Stoich. ratio of B to lim. reactant	-	σ^2	Variance, population ^a	-
θ	Angular position	rad	τ	Mean time between collisions	s
θ	Argument	rad	τ_n	Natural period ^b	s
θ_c	Critical angle	rad	τ_D	Rate time ^b	s
θ_p	Dead time ^b	s	τ	Reactor space time	s
θ	Fraction of gas feed permeated ^d	-	τ_I	Reset time ^b	s
θ_v	Molar conc. of vacant sites	$mol g_{cat}^{-1}$	τ_{xy}	Shear stress	Pa
θ_A	Sfc. conc. of sites occupied by A	$mol g_{cat}^{-1}$	τ	Shear stress tensor	Pa
θ	Zenith angle	rad	τ	Time constant	s
κ	Debye spring constant	-	$\vec{\tau}$	Torque	N m
κ	Isothermal compressibility	Pa^{-1}	$\bar{\tau}$	Tortuosity	-
κ	Torsion spring constant	$N m rad^{-1}$	Φ_Q	Flux density, heat	$W m^{-2}$

Φ_E	Flux, electric	V m
Φ_B	Flux, magnetic	Wb
Φ_G	Flux, mass, gas	$\text{kg s}^{-1} \text{m}^{-2}$
Φ_L	Flux, mass, liquid	$\text{kg s}^{-1} \text{m}^{-2}$
Φ_m	Flux, mass, relative to stationary coordinates	$\text{kg s}^{-1} \text{m}^{-2}$
Φ_n	Flux, molar, relative to stationary coordinates	$\text{mol s}^{-1} \text{m}^{-2}$
Φ_v	Viscous dissipation function	s^{-2}
ϕ	Azimuthal angle	rad
ϕ	Degree of inlet dilution ^g	-
ϕ	Fugacity coefficient	-
φ	Osmotic coefficient	-
ϕ	Phase angle	rad
ϕ	Phase shift	rad
ϕ	Solvent association factor ^d	-
ϕ	Void fraction	-
ϕ	Work function	eV
χ	Free moisture content	$\text{kg}_{(l)} \text{kg}_{\text{dry solid}}^{-1}$
χ_c	Free moisture content, crit.	$\text{kg}_{(l)} \text{kg}_{\text{dry solid}}^{-1}$
$\Omega_{D,AB}$	Collision integral, diffusive	-
Ω	Microcanonical number of states	-
Ω	Vorticity tensor	s^{-1}
ω	Frequency, angular	rad s^{-1}
ω_c	Frequency, cutoff	s^{-1}
ω_D	Frequency, Debye	s^{-1}
ω_L	Frequency, Larmor	rad s^{-1}
ω_0	Frequency, natural ^b	s^{-1}
ω	Pitzer acentric factor	-
$\vec{\omega}$	Velocity, angular	rad s^{-1}

Pseudoelement Symbols

A	Deprotonated acid
B	Brønsted–Lowry base
Et	Ethyl
HA	Brønsted–Lowry acid
M	Metal
Me	Methyl
Nu	Nucleophile
R	Functional group
X	Halogen

Superscripts

dep	Departure
E	Excess
(<i>n</i>)	<i>n</i> -th derivative / Lee-Kesler factor
$\alpha, \beta, \gamma, \dots$	Phase
*	Complex conjugate
‡	Transition state
<i>l</i>	Proper dimension / similar / other / predicted ^a
*	Pure species
o	Ideal
⊖	Standard / reference conditions
⊖	Steady state

Subscripts

a	Acid
b	Base
b	Basis ^e
bp	Bubble point
C	Cold
c	Critical

cat	Catalyst
comp	Compression
cv	Control volume
D	Bus load ^e
da	Drying agent / dry air
dp	Dew point
e	Equilibrium
eq	Equivalent
exp	Expansion
FL	Full load ^e
<i>f</i>	Final
fl	Flow
fp	Freezing point
G	Bus generator ^e
<i>G</i>	Gas stream
(g)	Gas phase
H	High
H	Hot
HK	Heavy key ^d
I	Inner
<i>i</i>	Index
<i>i</i>	Initial
<i>L</i>	Liquid stream
L	Low
Ld	Load ^e
LK	Light key ^d
Ln	Line ^e
(l)	Liquid phase
lm	Log-mean
max	Maximum / amplitude
min	Minimum
m:f	At interface from m to f
NL	No load ^e
<i>n'</i>	Moles of all other species held constant
O	Outer
<i>P</i>	At constant pressure (isobaric)
<i>P</i>	Permeate gas stream ^d
p	Particle
<i>Q</i>	At constant heat (adiabatic)
R	Receiving end ^e
<i>R</i>	Reject gas stream ^d
ref	Reference
rev	Reversible
rms	Root mean square
S	Sending end ^e
<i>S</i>	Solids
S	Source
<i>S</i>	Steam
s	Shaft
(s)	Solid phase
sat	Saturation
sfc	Surface
sh	Superheat
sp	Spring
<i>T</i>	At constant temperature (isothermal)
T	Total
Th	Thévenin
<i>V</i>	At constant volume (isochoric)
v	Valve
(v)	Vapour phase

W	Waste ^d
w	Water
wb	Wet bulb
μ	At constant chemical potential
ρ	At constant density (isopycnic)
0	Default / initial
δ	Earth

Sets, Symbols, and Diacritics*

\mathbb{C}	The set of all complex numbers
\hat{e}	Unit vector
\mathbf{I}	Identity matrix
i	Imaginary unit
j	Imaginary unit, electrical engineering
\mathbb{N}	The set of all natural numbers
\mathbb{N}^*	The set of all natural numbers and 0, <i>i.e.</i> $\mathbb{N} \cup \{0\}$
\hat{n}	Normal vector
\mathbb{O}^+	The set of all positive odd integers
\mathbb{Z}	The set of all integers
\varnothing	Diameter
\varnothing_h	Hydraulic diameter
$*$	Catalyst active site
\bullet	Intermediate
\parallel	Parallel
\perp	Perpendicular
\leftrightarrow	Horizontal
\updownarrow	Vertical
\perp	Bottom
\top	Top
$\{\square\}$	Numeric condition / set builder
$\langle \square \rangle$	Mean, population ^a / expected value / vector
$\bar{\square}$	Mean, sample ^a / partial molar / overall / below feed ^d / phasor ^e
$\bar{\bar{\square}}$	Grand average ^c
$\tilde{\square}$	Linear property (per length) / median, sample ^a
$\overset{*}{\square}$	Dimensionless quantity
$\underset{*}{\square}$	Temporal property (per time [<i>i.e.</i> rate]) / per unit ^e
$\overset{\square}{\square}$	Molar property (per mole)
$\hat{\square}$	Specific property (per mass) / estimator ^a
$\hat{\square}$	Areal property (per area)
$\underset{\square}{\square}$	Volumetric property (per volume)

Operators, Relations, and Functions

$\frac{D}{Dt}$	Material derivative
\det	Determinant
$E(t)$	Residence time distribution
erf	Error function
\mathcal{F}	Fourier transform
$H(s)$	Gain transfer function
\mathcal{H}	Hamiltonian
\Im	Imaginary part
\mathcal{L}	Laplace transform
\mathbb{P}	Probability
$R_{n\ell}$	Radial wave function
\Re	Real part
$t_{\alpha,\nu}$	t-statistic ^a
u	Unit step function
x_c	Complimentary solution
x_p	Particular solution
$Y_\ell^{m_\ell}$	Spherical harmonic
$y(x,t)$	Travelling wave
Γ	Gamma function

Δ	Difference / change
$\underline{\Delta}$	Relative change
δ	Dirac delta function (unit impulse)
δ	Inexact differential
δ_{ij}	Kronecker delta
ϵ_{ijk}	Levi-Civita symbol
Ψ	Wave function, time dependent
ψ	Wave function, time independent
φ	Probability density function ^a
$:$	Double dot product
$\nabla \times$	Curl
$\nabla \cdot$	Divergence
∇	Gradient
∇^2	Laplacian
\square^T	Transpose
\prod	Product
\sum	Sum
\wedge	Logical AND
\vee	Logical OR
\exists	Existential quantifier ("there exists")
\forall	Universal quantifier ("for all")
\cap	Intersection
\cup	Union
\subset	Is a subset of
\in	Is an element of
\equiv	Is defined as / defines
$[=]$	Has units of
\propto	Is proportional to
\Rightarrow	Implies / then
\rightarrow	Transform / next / proceed / approaches
\therefore	Because
\therefore	Therefore

Constants

a_0	Bohr radius	$5.292 \times 10^{-11} \text{ m}$
\mathcal{A}	Avogadro's number	$6.022 \times 10^{23} \text{ mol}^{-1}$
b	Wien's displacement constant	$2.897 \times 10^{-3} \text{ m K}$
c_0	Speed of light in vacuum	$299\,792\,458 \text{ m s}^{-1}$
e	Elementary charge	$1.602 \times 10^{-19} \text{ C}$
\mathcal{F}	Faraday's constant	$96\,485.33 \text{ C mol}^{-1}$
G	Gravitational constant	$6.674 \times 10^{-11} \text{ m}^3 \text{ kg}^{-1} \text{ s}^{-2}$
g	Local accel. of gravity, Earth	9.81 m s^{-2}
h	Planck constant	$6.626 \times 10^{-34} \text{ m}^2 \text{ kg s}^{-1}$
k_B	Boltzmann constant	$1.380 \times 10^{-23} \text{ J K}^{-1}$
k_e	Coulomb's constant	$8.987 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$
m_e	Mass of electron	$4.109 \times 10^{-31} \text{ kg}$
m_p	Mass of proton	$1.673 \times 10^{-27} \text{ kg}$
R	Gas constant, universal	$8.314 \text{ J mol}^{-1} \text{ K}^{-1}$
R_∞	Rydberg constant	$1.097 \times 10^7 \text{ m}^{-1}$
ϵ_0	Permittivity of free space	$8.854 \times 10^{-12} \text{ F m}^{-1}$
μ_B	Bohr magneton	$9.274 \times 10^{-24} \text{ J T}^{-1}$
μ_0	Permeability of free space	$4\pi \times 10^{-7} \text{ H m}^{-1}$
σ	Stefan-Boltzmann constant	$5.676 \times 10^{-8} \text{ J s}^{-1} \text{ m}^{-2} \text{ K}^{-4}$

Notes

- a* Linear regression / ANOVA / statistics
- b* Process control
- c* Statistical process control
- d* Distillation / separation
- e* Power engineering
- f* Mechanical engineering / materials science
- g* Process safety
- ** Specific definitions for symbols with diacritics take precedence

